

Physico-chemical Characteristics of Sokoto Locally grown *Cucumis melo* L (Honeydew Melon) Seed Oil

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Abstract

Introduction: Owing to increase demand for safer and health promoting vegetable oils, a number of potential sources are being explored by researchers.

Materials and Methods: In this study, oil was extracted using Soxhlet from Sokoto locally grown *Cucumis melo* L (honeydew melon). Physical and chemical properties (colour, moisture, pH, specific gravity, refractive index, acid value, iodine value, saponification value and peroxide value) of the oil were determined using standard analytical methods by Association of Official Analytical Chemists (AOAC).

Results: The results showed the percentage yield of the oil to be 27.46%. Physically, the oil was yellowish in colour, liquid at room temperature, with pH of 6.2 (0.01), specific gravity of 0.89 (0.32), and refractive index of 1.362 (1.0). The saponification, acid, iodine and peroxide values of the oil were 45.81 (5.19), 9.16 (0.21), 64.80 (4.31) and 10.50 (1.50) respectively.

Conclusion: The results suggest that the oil has a potential for use as vegetable oil, in industries and, subject to further evaluation of the contents, health promoting purposes.

Key words: honeydew melon, oil, physico-chemical characteristics

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Introduction

There has been an emerging interest in the compounds that occur naturally in plants and their extracts owing to their therapeutic and risk minimizing potentials for many chronic diseases such as cardiovascular diseases, cancers, diabetes and neurodegenerative diseases like Alzheimers' disease (1,2). Honeydew melon (*Cucumis melo*) (Figure 1A) also called sweet melon or golden melon is grown in temperate climatic regions of the world including Nigeria (in West Africa) in which Sokoto state is located. Its wide spread cultivation is as a result of the taste of the fruit and applications in traditional medicine (3). It has been used for cardiovascular diseases, liver and kidney diseases, anaemia, rheumatism and gout (4). The seeds of honeydew melon (Figure 1B) in some parts of the world are regarded as waste, not until some researchers began to explore their benefits such as the use as antioxidants, anti-inflammatory and analgesics (5).

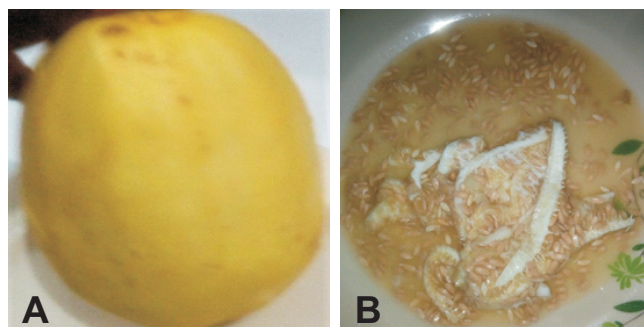


Figure 1. *Cucumis melo* L (Honeydew melon) A= Fruit, B= Seeds

Hitherto, there appeared to be few reports on the extraction of oil from the seeds and characterisation of the oil especially of the African variety. The reports in other parts of the world showed that the honeydew seed oil is rich in tocopherols, sterols, and phospholipids (6). However, the variation of compositions of the oil depending on the type of the melon and on the area it originates was noted by some researchers (6,7,8). In

Africa, there is increasing demand for vegetable oils commensurate with the rapid population growth which, in a way, overwhelm the few sources available. Hence, there is need to explore cheaper, safer and nutritionally rich vegetable oils. In the present study, oil was extracted from locally grown honeydew melon seeds and some physical and chemical properties were determined.

Materials and Methods

Materials

The honeydew melon fruit was purchased from Gawon nama local market Sokoto. The seeds were harvested from the fruits, dried for 72 hours at 25°C and then ground to powder using clean mortar and pestle. The chemical used were of analytical grade. Glass Soxhlet extractor was used for the oil extraction.

Extraction of oil

Oil was extracted from honeydew melon seed by Soxhlet using n-hexane as the solvent according to AOAC method (9). Fifty grams (50 g) of shelled ground seeds were packed into a weighed thimble, which was introduced into the Soxhlet system. About 250 mL of the solvent (n-Hexane) was poured into the flask. The set up was damped and heated on a heating mantle. The extraction processes were carried out in six hours at temperature of n-Hexane (60°C). Solution obtained from the extraction was regained through a reflux process for 15 minutes after which the content was evaporated by increasing the temperature to 70°C for about half an hour to obtain yellowish, liquid at room temperature oil. The pH of the oil was determined using digital bench top pH meter (pH/mV/ISE) and was then subjected to further analysis.

Determination of physical characteristics honeydew melon seed oil

Determination of moisture content

The moisture content of the oil was determined by air-oven method, according to Ghatak and Panchal (10) with slight modification. Five grams (5g) of oil were weighed in previously dried, weighed and tared dish. The dish was heated for 1 hour at about 105°C. The dish was then removed, cooled in a desiccator and then weighed. The dish was heated again for another 1 hour, then was allowed to cool down before being weighed. This was repeated until the change in weight between observations was less than 1mg. The moisture content (percentage by

weight) was calculated by
$$\frac{W1}{W} \times 100$$

Where W1 = weight loss (g) of the sample on heating

W = weight of the sample (g) taken

Determination of specific gravity

Specific gravity of the oil was determined according to the method adopted by Ghatak and Panchal (2010) using a specific gravity bottle. The bottle was filled with the oil sample, avoiding entrapment of air bubbles from the side arm, stopper was replaced and then immersed in the water bath at 25°C and held for 30 minutes. The bottle was then removed from the water bath and dried. The cap of the side arm was removed and the bottle quickly weighed. This was repeated using water instead of the oil.

The specific gravity of the oil was calculated by

$$\text{Specific gravity at } 25^{\circ}\text{C} = \frac{A-B}{C-B}$$

Where A= weight in gram of specific gravity bottle with oil at 25°C

B= weight in gram of specific gravity bottle at 25°C

C= weight in gram of specific gravity bottle with water at 25°C

Determination of refractive index

The refractive index of the oil was determined using Abbe refractometer at 25°C (AOAC 921.08). Few drops of the oil were placed, ensuring complete cover and avoiding bubbles, on the polished surface of the measuring prism. The illuminating prism was brought into contact with the measuring prism and reading was taken. The procedure was repeated thrice and average and standard deviation of the readings were taken.

Determination of chemical characteristics of honeydew melon seed oil

Determination of saponification value

Saponification value of the oil was determined according to AOAC method (1976). Four grams (4g) of oil was weighed into a flask. From burette, 50mL of alcoholic KOH was added and it was allowed to drain for a definite period of time. The blank was prepared by taking 50mL of alcoholic KOH which was allowed to drain at the same duration of time. Air condenser was connected to the flasks and they were gently boiled for about 1 hour. When the flask and condenser got cooled, the condenser was

rinsed using distilled water (just a little of it). It was then removed and 1mL of the indicator was added. The mixture was then titrated against 0.5N HCL until the pink colouration disappeared.

The saponification value was calculated from the titre value as follows:

$$\text{Saponification value} = \frac{28.05 \times (\text{titre value of blank} - \text{titre value of sample})}{\text{Weight of sample (g)}}$$

Determination of Iodine value

Iodine value of the oil was determined according to the AOAC (1996) method. The oil sample (250 mg) was weighed into an Iodine flask and 10mL of chloroform was dissolved. 25mL of iodine solution was added using a pipette. It was drained in a definite time, well mixed, and then allowed to stand in a dark cupboard for about 30 mins with occasional shaking. 10 mL of 15% KI was added and shaken thoroughly. Freshly boiled and cooled down water (100 mL) was added. Free Iodine on the stopper was washed down. The mixture was then titrated against 0.1 N sodium thiosulphate until the yellow solution turned colourless. A few drops of 1% starch (as indicator) was added and again, it was titrated until the blue colour disappeared completely. Iodine value was calculated by

$$IV = \frac{12.69 \times N(V_2 - V_1)}{W}$$

Where N = normality of sodium thiosulphate, V₂= volume (mL) of sodium thiosulphate for blank, V₁= volume (mL) of sodium thiosulphate consumed by sample and W= weight (g) of sample, 12.69 is a constant based on grams of iodine in 0.1 mol iodine solution.

Determination of peroxide value

Peroxide value of the oil was determined according to AOAC (1990). One gram (1 g) of the oil was weighed into a clean dry tube and 1 g of potassium iodide was added to 20 mL of solvent mixture. The tube was placed into boiled water, the liquid boiled rigorously within 30 seconds. The content was transferred quickly into a conical flask which already contained 20 mL of 5% potassium iodide solution. Each time the tube was washed twice with 25mL of water and collected into the conical flask. It was then titrated against 0.1 M sodium thiosulphate solution until the yellow colour disappeared. 0.5 mL of 1% starch solution was added. It was shaken vigorously and then titrated carefully until the blue colouration disappeared. A blank without the oil

was also determined. PV (meq/Kg) was calculated as follows

$$PV (\text{meq/Kg}) = \frac{M \times (V - V_0) \times 1000}{\text{Weight of sample (g)}}$$

Hence M is the molar concentration of sodium thiosulphate, V is the volume of titrant for sample and V₀ volume of titrant for blank.

Statistical analysis

Results were analysed using Microsoft excel 2016. Data were presented as mean (SD).

Results and Discussion

The results for physical characteristics of honeydew melon seed oil are as depicted in table 1. The colour of the oil was found to be yellowish which often indicates the presence of carotene (11). Colour often indicates an aesthetic quality of oil and guides bleaching and deodorization processes (12). Most oils are sold based on their colour and each type has sell by colour instructions.

Table Physical properties of honeydew melon seed oil at room temperature

Parameters	Value
Colour	Yellow
Physical state at room temperature	Liquid
% yield	27.49 (0.19)
% moisture	2.40 (0.91)
Ph	6.21 (0.01)
Specific gravity (g/cm ³)	0.88 (0.02)
Refractive index	3.62 (0.9)

Data presented as mean (SD)

The physical status of the oil was liquid at room temperature. This indicates the presence of, predominantly, unsaturated fatty acids which characteristically kink to prevent packing together, thus making the oil liquid. Oils containing high amount of unsaturated fatty acids have been shown to confer protection against the risk of cardio-vascular diseases (13,14). The percentage yield of oil 27.49% was lower than previously reported 44.89% (8), 41.6% (15); and was higher than what was reported by Yanty et al. (16). The variation could largely be due to the variety of the melons used, solvent used, extraction time, and to some extent the methods of oil extraction. For Yanty et al. (16) used petroleum ether, while n-Hexane was used in this study and this could explain the reason for the variation. The variation in oil yield of this study and that of Petkova and Antova (15) is likely due to extraction times (6 vs 8

hours respectively). However, variety of the melons could account for the yield variation of this study (*Cucumis melo L*) with that of Ibeto et al. (*Cucumis melo Cylindrica*) (8).

Lower moisture content has been reported to increase oil recovery and decrease oil loss during cooking (17). The refractive index of oils tells about their purity, molecular weight, fatty acid chain length, degree of un-saturation, and degree of conjugation (18). Honeydew melon seed oil has a refractive index higher than that of water whose refractive index is 1.334. The refractive index of an oil is dependent on its density and reflects fatty acid chain length, degree of unsaturation and triglycerides contents of the oil (8). Values of refractive index increases with increase in the levels of triglycerides, because triglycerides have higher refractive indices than free fatty acids (19). Yanti and colleagues (16) reported high triglycerides in melon seed oil with predominant linoleic acid. Hence, the high refractive index obtained in this study could be due to high triglycerides content of the oil, however, triglycerides were not determined. It is important to note that, triglycerides serve as primary energy stores in the body and provide fatty acids for phospholipids biosynthesis which are important constituents of cell membrane. However, excessive triglyceride has been reported as an independent risk to development of cardiovascular disease (22).

The results of the chemical characteristics of melon seed oil are as depicted in table 2. Saponification value, a measure of tendency to soap formation and difficult separation of oils, indicates the presence of high amount of fatty acids (20). The saponification value of the oil in this study is lower than most reported previously, thus the oil may not be suitable for soap making, however, could be utilized for biodiesel formation, since low saponification value allows for separation (8).

Table 2 Chemical characteristics of honeydew melon seed oil at room temperature

Parameters	Value
Saponification value (mg KOH/g)	45.81 (5.19)
Acid value (mg KOH/g)	9.16 (0.21)
Iodine value (gI ₂ /100g)	64.80 (4.31)
Peroxide value (meq/Kg)	10.50 (1.50)

Data presented as mean (SD)

Acid value an important measure of rancidity of oils, usually formed from hydrolysis of triglycerides was 9.6(0.21) mgKOH/g. This value is above the permissible level of less than 1, thus the oil has high potential of

rancidity within short time. However, the high acid value could be as a result of high triglycerides suggested by high refractive index.

The Iodine value obtained for honeydew melon seed oil, which measures the degree of unsaturation was lower than the ranges for different oils reported by previous studies (8,10,15,16,21). However, it is within FAO/WHO acceptable limits of 50-129 gI₂/100g. iodine value influences long-term stability properties of oil which is important for storage. The peroxide value, a measure of oxidation products of oils was slightly higher than previously reported (8). However, this may not be unconnected to the presumably high levels of triglycerides.

Conclusion

This study revealed that honeydew melon seed is a promising source of healthy vegetable oil that is fairly stable oxidatively and has potential for use in industries and production of biodiesels. The health potential of this oil is high if further explored for the presumptive triglyceride contents. The Iodine value being within acceptable limit also confers potential long-term stability to the oil, a good factor for storage. However, the oil may be highly rancid due to the high acid value. Studies are underway for fatty acids profile and bioactive components; and possible testing of the bioactive components of the oil.

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